

Unusual Intermolecular Ligand Exchange Process in Octakis(trimethyl phosphite)dirhodium(0), $\text{Rh}_2[\text{P}(\text{OMe})_3]_8$

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Summary. The new complex $\text{Rh}_2[\text{P}(\text{OMe})_3]_8$ which has a bicapped trigonal antiprismatic structure at room temperature undergoes a specific equatorial intermolecular ligand exchange at higher temperatures rather than any intramolecular ligand exchange between equatorial and axial sites.

THERE is current interest in the fluxional behaviour of five-co-ordinate transition-metal complexes.¹⁻⁴ We report the preparation of the pale yellow crystalline complex octakis(trimethyl phosphite)dirhodium(0) $\text{Rh}_2[\text{P}(\text{OMe})_3]_8$ (I), m.p. 75° (decomp.), (60%) by displacement of the cyclopentadienyl group and ethylene from $\pi\text{-C}_5\text{H}_5\text{Rh}(\text{C}_2\text{H}_4)_2$ with excess of trimethyl phosphite at 60°.

The proton-decoupled ³¹P n.m.r. spectrum of (I) [Figure (a)] at room temperature is of the A_3BX type (A and B = ³¹P nuclei, X = ¹⁰³Rh, 100% natural abundance) indicating that (I) has a non-fluxional symmetrical bicapped trigonal antiprismatic structure (each rhodium having local C_{3v} symmetry). This is similar to that reported very recently for the non-fluxional cobalt complex $\text{Co}_2[\text{P}(\text{OMe})_3]_8$.⁴ Analysis of the ³¹P n.m.r. spectrum yields: J_{AB} 122, J_{AX} 200, $J_{BX} \approx J_{AB} = 122$ Hz.

The temperature dependence of the ³¹P n.m.r. spectrum of (II) has been studied between room temperature and 100°. Both sets of signals broaden around 50° and near 100° the A part of the spectrum becomes a sharp singlet [Figure (c)] while the B part appears as a doublet separated by 122 Hz (J_{BX}). There is no change in the chemical shift difference $\delta_A - \delta_B$ over the temperature range and the original spectrum is regained on cooling the sample to room temperature.

These results indicate that the molecule is *not* undergoing an intramolecular exchange of phosphite ligands and the loss of rhodium-phosphorus coupling only for resonance A implies that there is a specific intermolecular ligand exchange in the equatorial positions at higher temperatures. The lack of intermolecular ligand exchange at room temperature is confirmed by adding a small amount of free trimethyl phosphite to a solution of (I). The resulting room temperature ³¹P n.m.r. spectrum [Figure (b)] consists of a simple superposition of the spectrum of (I) and that of the free phosphite. As the temperature is raised to 100° there is a coalescence of the A part of the spectrum and the single line of the free ligand to give a spectrum similar to the Figure (c). The B part remains as a 1-1 doublet owing to coupling with rhodium.

This appears to be the first example of a five-co-ordinate complex undergoing a reversible intermolecular ligand

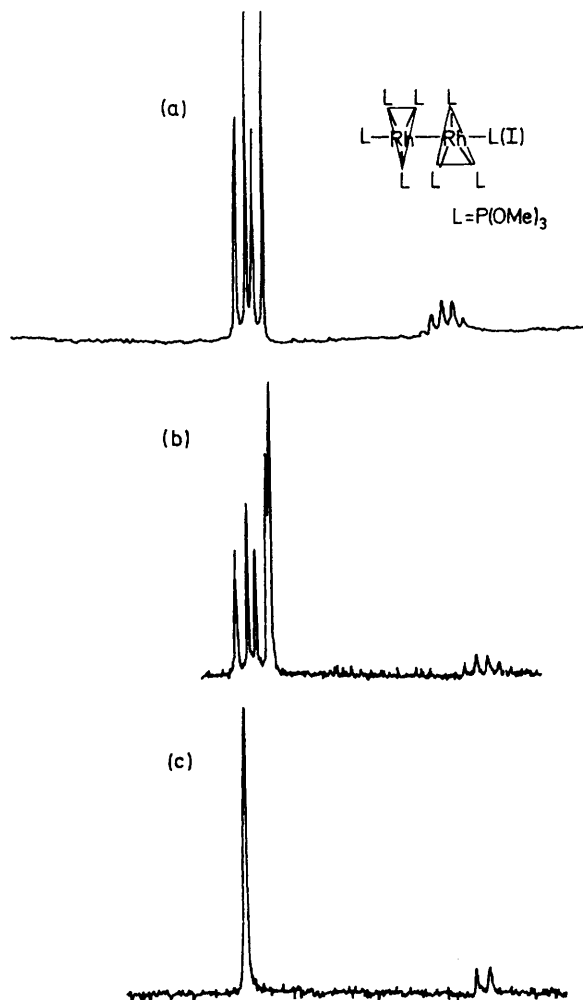


FIGURE. 40.49 MHz ³¹P {H} n.m.r. spectrum of $\text{Rh}_2[\text{P}(\text{OMe})_3]_8$: (a) at 32°; (b) with added $\text{P}(\text{OMe})_3$ at 32°; (c) at 100°.

exchange specifically with the equatorial set of ligands rather than an intramolecular exchange process between equatorial and axial sites where the barriers are usually low (5-15 kcal/mol).

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